

## ${}^1_0\text{H}$ , ${}^2_0\text{H}$ , ${}^3_0\text{H}$ (Isotope)

${}^1_0\text{H}$ ,  ${}^2_0\text{H}$ ,  ${}^3_0\text{H}$  are isotopes of hydrogen.  ${}^1_0\text{H}$  is the most common,  ${}^2_0\text{H}$  is deuterium, and  ${}^3_0\text{H}$  is tritium.

${}^1_0\text{H}$  is stable,  ${}^2_0\text{H}$  is stable, and  ${}^3_0\text{H}$  is radioactive.

${}^1_0\text{H}$  is used in many applications,  ${}^2_0\text{H}$  is used in heavy water, and  ${}^3_0\text{H}$  is used in nuclear weapons.

${}^1_0\text{H}$  is the most common isotope of hydrogen,  ${}^2_0\text{H}$  is deuterium, and  ${}^3_0\text{H}$  is tritium.

${}^1_0\text{H}$  is stable,  ${}^2_0\text{H}$  is stable, and  ${}^3_0\text{H}$  is radioactive.  ${}^1_0\text{H}$  is used in many applications,  ${}^2_0\text{H}$  is used in heavy water, and  ${}^3_0\text{H}$  is used in nuclear weapons.